

Quality Assessment of Some Freshwater Resources Located in Bucharest and Surrounding Areas

II. Water quality assessment of Arges and Dambovită rivers

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The water quality assessment of Arges River and its tributary Dambovită River is presented in this paper as second part of an extended study on the evaluation of some freshwater resources quality located in Bucharest and surrounding areas [1]. This case study was carried out mainly for the water in the confluence area of Arges and Dambovită rivers, where a water and sediment sampling campaign was organized in June 2018. 21 quality parameters were determined for water samples and for the sediment samples the heavy metals content was evaluated. The results of the water samples analysis allowed the framing in quality classes, and for sediment samples the values obtained were compared with the chemical quality standards for sediments, according to the current national regulations. Following the assessment, it was established that the freshwaters in both Argeș River and Dambovită River, downstream the confluence with the Dambovită River are loaded with different pollutants, such as organic substances and nutrients leading to a lower water quality classification.

Keywords: freshwater, water quality, pollution, Arges River, Dambovită River

Freshwater is the most important natural resource for life and the environment, nevertheless it is a vulnerable and limited resource, especially due to its irrational exploitation [2-4]. The availability of water in sufficient quantities and of good quality will become even more challenging for Europe in the future due to climate change, the growing needs of the urban population, as well as, industry and agriculture expansion [5-8].

Nowadays, the protection of aquatic life and human life is a permanent concern, so monitoring the quality of surface water has an important role to play [9, 10]. According to studies in the field, the presence of chemical pollutants in the environment, especially from anthropogenic sources, represents a threat to living organisms [11-13]. Pollutants and potentially toxic elements (like heavy metals) can accumulate in quantities exceeding the maximum admissible limits in both surface and underground waters and soil. The negative effects of pollutants can be expressed both directly on the organisms that populate the respective aquatic basin and indirectly by diminishing oxygenation, changing the pH and other physical and chemical properties of the water [4]. Among the harmful consequences of these substances we can mention the carcinogenic and mutagenic effects, accumulation in the food chain links, high toxicity etc., all of which contributing to the serious disturbance of the natural equilibrium [14-17].

In urban areas, the quality of the aquatic ecosystem is considerably influenced by industrial and agricultural effluents, as well as wastewater collected in the sewage system [18]. In Romania, at the Bucharest Municipality level, the water of Dambovită River is mainly used as drinking water supply. The Dambovită River has a length of 286 km and is the largest tributary of the Arges River (350 km) [19]. Bucharest Municipality is the main city having

significant impact on the two rivers, because a part of the wastewater collected through the sewerage system is discharged into the Dambovită River downstream of its Bucharest course where the sewage treatment plant is located (at Glina point), but due to the wastewater treatment plant incapacity to take over the total collected wastewater volume, the natural regime of the Dambovită River is sensitively modified by the urban influence [20]. Also, the rapid and unplanned expansion of the constructions in the surrounding areas of Bucharest city has caused water supply problems related to the lack of efficient sewerage systems and the emergence of uncontrolled landfills which had a significant impact on the freshwater resources quality. As a consequence, an extended study on evaluation of the freshwater resources quality in some Bucharest and surrounding areas was performed in order to establish the freshwater quality classes in accordance to the provisions of the European Water Framework Directive (WFD) 2000/60/EC [21] which has been implemented in Romanian regulations [22].

The Part I of this work related a case study on water quality assessment of the Mogosoia, Herastrau and Pantelimon Lakes, built on the Colentina River which is the tributary of the Dambovită River [1]. This paper is the Part II which presents the water quality assessment of Arges and Dambovită rivers at their confluence area through the physicochemical parameters according to national current regulations.

Experimental part

Study area and samples collection

The selected area for this study is situated at the confluence of the Arges River with its tributary, the Dambovită River, included in the Arges-Vedea hydrographic area, south Romania. To achieve the study objective, a

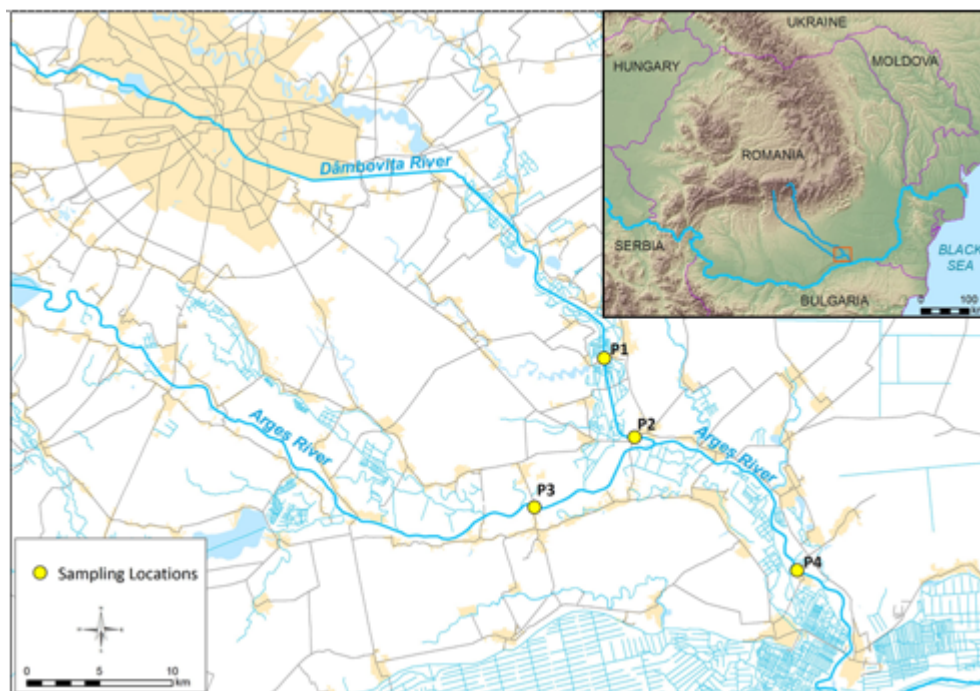


Fig. 1. Location of the study area

Notation of sampling site locations		Geographical coordinates	
		Latitude	Longitude
Dambovitza/Nuci	P1	4428116°N	2642888°E
Dambovitza/Budesti	P2	4423198°N	2645384°E
Arges/Hotarele	P3	4418984°N	2636657°E
Arges/Clatesti	P4	4414777°N	2659128°E

Table 1
SAMPLING SITE LOCATIONS

water and sediment sampling campaign was organized in June 2018 on the two mentioned rivers. Figure 1 shows the study area location and Table 1 shows the four sampling locations P1-P4 (P1-Dambovitza/Nuci, P2-Dambovitza/Budesti, P3-Arges/Hotarele, P4-Arges/Clatesti) with their geographical coordinates.

The water samples were collected in polyethylene recipients (3 L) from approximately 30 cm below water surface and were kept at 4 °C during their transport to the laboratory, according to the in force standards [23-25]. Sediment samples were collected according to sampling procedure of the current standards [26]. *In situ* measurements were performed for the determination of unstable parameters: temperature, pH, conductivity and dissolved oxygen using a portable multi-parameter WTW Multi 340i. The device was calibrated using standard solutions before each determination.

The determination of quality parameters

The following 21 physicochemical parameters were determined for water samples: temperature (T), pH, electrical conductivity (EC), turbidity, dissolved oxygen (DO), chemical oxygen demand (COD_C and COD_{Mn}), 5-days biochemical oxygen demand (BOD₅), total dissolved solids (TDS), chloride (Cl⁻), sulphates (SO₄²⁻), ammonium-nitrogen (NH₄⁺-N), nitrate-nitrogen (NO₃⁻-N), nitrite-nitrogen (NO₂⁻-N), total nitrogen (TN), total phosphorus (TP), orthophosphates (PO₄³⁻-P), Chlorophyll a (Chl a), cadmium (Cd), lead (Pb) and copper (Cu).

After reception of the sediments samples in the analysis laboratory, they were air dried at room temperature. After drying, to obtain representative sediments samples, they were milled, sieved and in order to bring them into the

solution, mineralization of about 0.5 g of the sample from the 63 μ m fraction in the presence of aqua regia has been performed. For the assessment of the sediment quality, the following heavy metals have been determined: cadmium (Cd), mercury (Hg), copper (Cu), nickel (Ni), lead (Pb), chromium (Cr) and zinc (Zn). For heavy metals determination a Atomic Absorption Spectrophotometer - ContrAA 700 (Analytikjena) have been used.

All reagents used for the determination of the physicochemical parameters were of analytical purity and the analytical determinations have been performed using standardized methods of analysis.

Results and discussions

Water quality assessment

The assessment of the water samples quality collected from sampling locations P1-P4 was carried out following the national regulations, according to which the physical and chemical indicators are grouped as follows: thermal regime and acidification, oxygen regime, nutrients, salinity, specific toxic pollutants of natural origin and other relevant chemical indicators [22].

According to the performed analyses, the pH values were within the limits specified by M.O. 161/2006 [22] and ranged from 7.36 to 7.99 pH units and the temperature varied within the range 24.5 - 27 °C.

Figure 2 shows the water quality classes for the geo-referenced *oxygen regime* indicators at the sampling locations of the water samples according to the mentioned order. For DO parameter, the minimum value of 0.8 mgO₂/L was determined in P1 and the maximum value of 8.00 mgO₂/L was determined in P3. The COD_C values determined varied in the range 8.77 mgO₂/L (P3) - 50.42

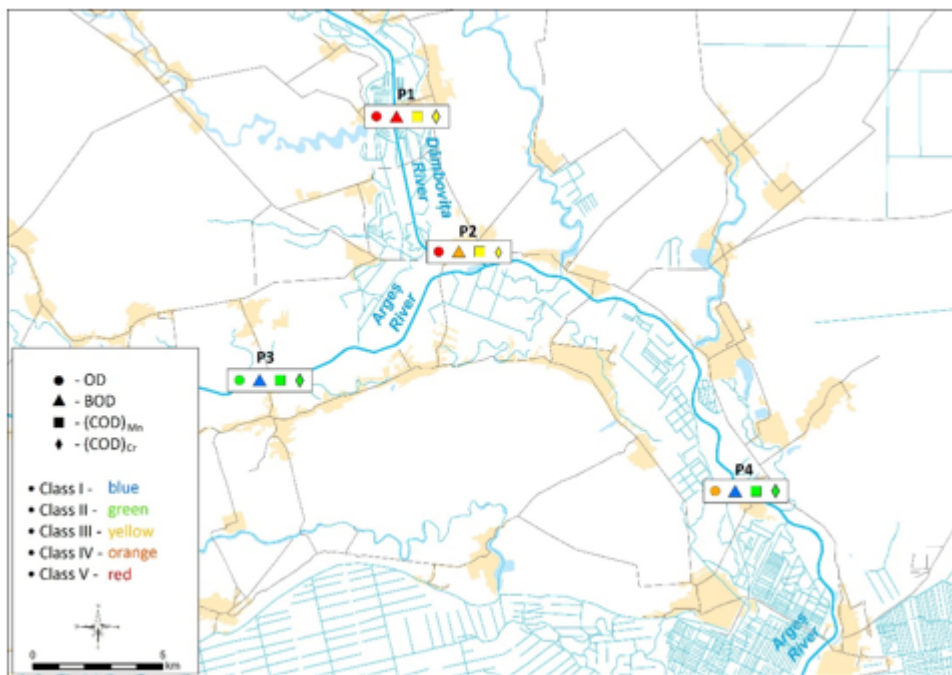


Fig. 2. Water quality classes for the indicators DO, COD_{Cr} , COD_{Mn} and BOD geo-referenced for each sampling location

mgO_2/L (P1); COD_{Mn} values varied between $6.94 mgO_2/L$ (P3) - $19.95 mgO_2/L$ (P1) and BOD values recorded a minimum of $2.06 mgO_2/L$ (P3) and a maximum of $31.02 mgO_2/L$ (P1).

From figure 2 it can be noticed that in the case of two sampling locations P1 and P2 established for the Dambovitza River the DO values have been included in the water quality Class V and the COD (COD_{Cr} and COD_{Mn}) values in water quality Class III. The BOD values obtained were classified as quality Class V for P1 and Class IV for P2. In the case of sampling locations established for the Arges River (P3 and P4), the DO values were included in water quality Class II for P3 and IV for P4, the BOD values were included in water quality Class I for the both locations and also the COD_{Cr} and COD_{Mn} values were included in water quality Class II for both locations.

Figure 3 shows the water quality classes according to their values obtained for the quality indicators of *nutrients regime*. The values obtained for NH_4-N ranged from $0.10 mg/L$ (P3) to $9.17 mg/L$ (P1), the NO_3-N values determined in the four sampling locations ranged from $0.03 mg/L$ (P2) to $0.49 mg/L$ (P3) and the TN values varied between 1.49

mg/L (P3) and $10.49 mg/L$ (P1). Regarding the values recorded for PO_4-P and TP, these were in the range of $0.05 mg/L$ (P2) - $0.73 mg/L$ (P1), respectively $0.22 mg/L$ (P3) - $1.20 mg/L$ (P1). The values obtained for Chlorophyll *a* ranged from $2.37 mg/L$ (P3) to $15.04 mg/L$ (P1).

From figure 3 it can be seen that for P1 and P2 sampling locations on Dambovitza River, the values obtained for the quality indicators of the nutrients regime have led to similar quality classes for the two selected locations. Thus, for both sections, the NH_4-N recorded values were specific to water quality Class V, NO_2-N values to quality Class III, TP values to quality Class II and the NO_3-N and Chlorophyll *a* values to water quality Class I.

In the case of water samples collected from P3 and P4, locations situated on the Arges River, values belonging to water quality Class I for NO_3-N , TN, Chlorophyll *a* were observed in both locations, values specific to the water quality Class IV for NH_4-N and NO_2-N in P4, values for quality Class III for NO_2-N and water quality Class I for NH_4-N in P3. The ammonium ion presence in a higher amount in the analysed samples from P4 may be due to the organic substances decomposition under anaerobic conditions and



Fig. 3. Water quality classes derived from nutrients geo-referenced for each sampling location



Fig. 4. Water quality classes for Salinity geo-referenced for each sampling location

in the presence of bacteria or reduction of nitrate ions. For locations situated on the Arges River, the TP and PO₄-P values obtained in P3 were specific to water quality Class I and for P4 were specific to quality Class I and Class II.

Regarding the *salinity* specific quality indicators, EC values ranged from 704 $\mu\text{S}/\text{cm}$ (P1) to 355 $\mu\text{S}/\text{cm}$ (P3), TDS values from 266 mg/L (P3) to 528 mg/L (P1), Cl values varied in the range of 17.5 mg/L (P3) - 63 mg/L (P1) and SO₄²⁻ values ranged from 37.61 mg/L (P3) to 45.57 mg/L (P1). The high values of conductivity are an indicator of high concentrations of dissolved ions in water. Figure 4 shows the water quality classes for TDS, Cl⁻ and SO₄²⁻.

From figure 4 it can be noticed that for the sampling locations situated on the Dambovitza River, the obtained values were included in similar water quality classes for both locations, as follows: the TDS values were specific to the quality Class II of water, the Cl values were specific to quality Class III and SO₄²⁻ values to quality Class I. In the case of samples collected from Arges River, the TDS and SO₄²⁻ values were in the water quality Class I for both locations and the Cl values were in the quality Class I in P3 and quality Class II in P4.

For the water samples collected from the four monitoring locations, the following heavy metals have been analysed: Cd, Pb and Cu. After evaluation of the values obtained we observed the following aspects: the Cd values ranged between 0.04 g/L in P1 and to below the method detection limit, respectively 0.02 mg/L for P2, P3 and P4, the Pb values ranged between 1.98 g/L in P1 and to below the method detection limit, respectively 0.30 $\mu\text{g}/\text{L}$ in P3 and Cu values varied between 4.88 $\mu\text{g}/\text{L}$ in P1 to 1.70 $\mu\text{g}/\text{L}$ in P4.

Figure 5 shows the variation of the heavy metal concentrations determined in water samples. Following the reporting of the heavy metals values obtained to the current national regulations, the classification in water quality Class I for all the values has been obtained.

Assessment of sediment quality

For the sediment quality assessment the following heavy metals have been analysed: Cd, Hg, Cr, Cu, Ni, Zn and Pb. Figures 6-9 show the variation of heavy metal concentration (mg/kg) in the sediment samples for each sampling locations.

After the assessment of heavy metal concentrations the values obtained for the determined heavy metals have

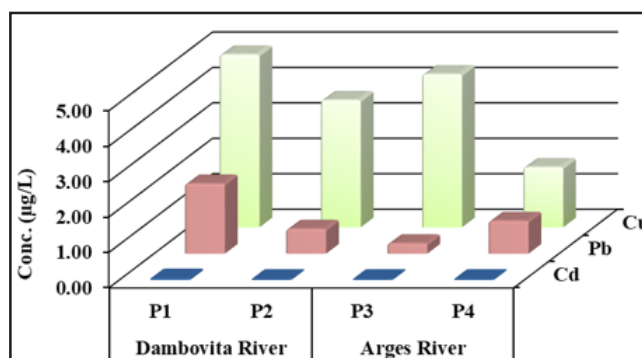


Fig. 5. The concentration of heavy metals Cd, Pb and Cu determined in water samples

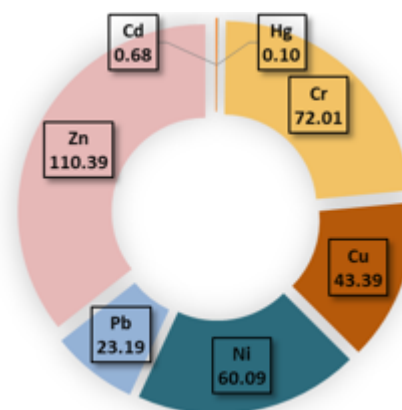


Fig. 6. The concentration of heavy metals in sediment samples in P1

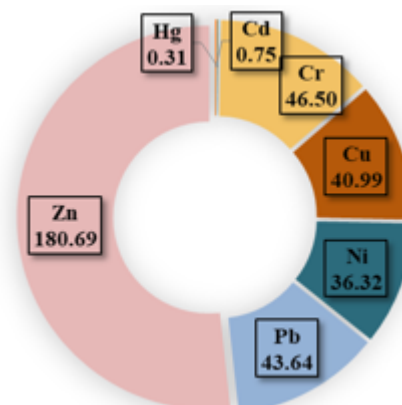


Fig. 7. The concentration of heavy metals in sediment samples in P2

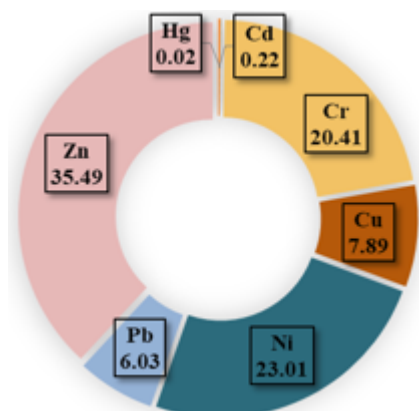


Fig. 8. The concentration of heavy metals in sediment samples in P3

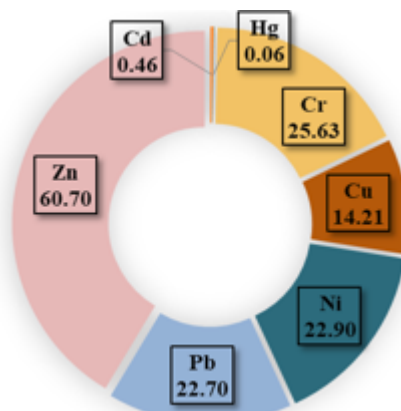


Fig. 9. The concentration of heavy metals in sediment samples in P4

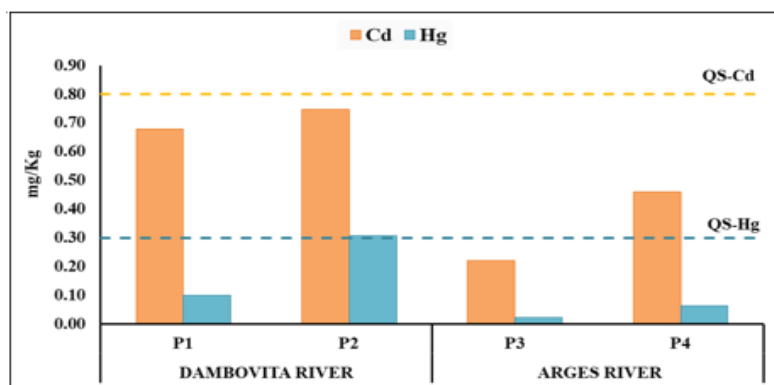


Fig. 10. The concentration of Cd and Hg in sediment samples compared to chemical quality standards

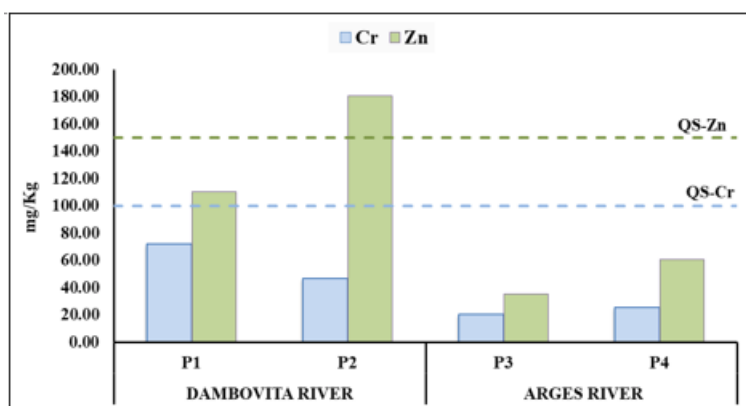


Fig. 11. The concentration of Cr and Zn in sediment samples compared to chemical quality standards

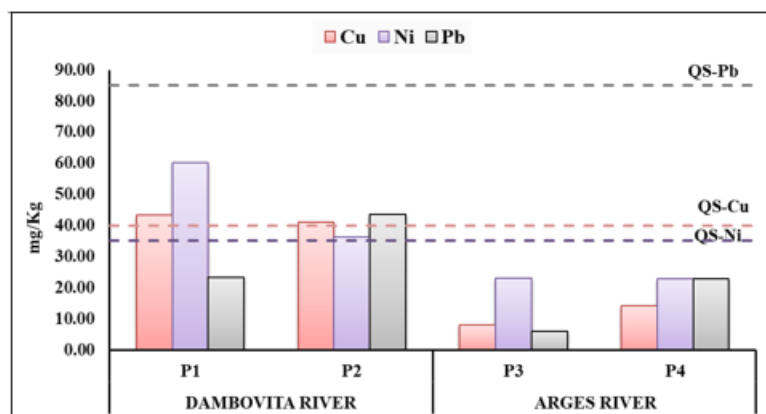


Fig. 12. The concentration of Cu, Ni and Pb in sediment samples compared to chemical quality standards

varied in order: $Zn > Cr > Ni > Cu > Pb > Cd > Hg$ in P1 (fig. 6), $Zn > Cr > Pb > Cu > Ni > Cd > Hg$ in P2 (fig. 7), $Zn > Ni > Cr > Cu > Pb > Cd > Hg$ in P3 (fig. 8) and $Zn > Cr > Ni > Pb > Cu > Cd > Hg$ in P4 (fig. 9). It has been observed that in P1, P3 and P4, heavy metals Zn, Cr, Ni showed a high concentration, and in P2, Zn, Cr and Pb recorded high values. Figures 10-12 show the framing of heavy metal concentrations determined in the sediment samples in chemical quality standards (QS) according to national regulations [22].

From analyzing the graphs presented above in figures 10-12, the values for Cd and Hg were highlighted below

the limits of the chemical quality standards, respectively $QS-Cd = 0.8 \text{ mg/Kg}$ and $QS-Hg = 0.3 \text{ mg/Kg}$ (fig. 10). Also, the values obtained for Cr were below the chemical quality standards ($QS-Cr = 100 \text{ mg/Kg}$) while Zn recorded slightly higher values only in P2 compared to the quality standard ($QS-Zn = 150 \text{ mg/Kg}$) (fig. 11). The values obtained for Pb were below the chemical quality standards ($QS-Pb = 85 \text{ mg/Kg}$), while the concentrations obtained for Cu and Ni recorded slight exceedances in P1 and P2 compared to the reference standards ($QS-Cu = 40 \text{ mg/Kg}$; $QS-Ni = 35 \text{ mg/Kg}$) (fig. 12).

Conclusions

The quality of freshwater resources, respectively of the Arges and Dambovită Rivers, has been assessed at their confluence area using physicochemical parameters according to the national regulations. Following the analyses carried out for the water samples collected from the two sampling locations on the Dambovită River, specific values of water quality Class V for organic substances and nutrients have been highlighted at locations Dambovită/Nuci and Dambovită/Budești. Also, for the sediment samples slight exceedances have been highlighted according to the chemical quality standards for Cu, Ni in location Dambovită/Nuci and Ni, Zn in location Dambovită/Budești, with the possibility of developing some minor pressure from these indicators on the studied freshwater resources.

Regarding the Arges River, the analyses revealed that from the physicochemical point of view, downstream of Budești after the river receives the polluted waters of the Dambovită River at Clatești, the water quality changes to quality Class IV, due to the presence of organic substances and nutrients. Pollution by organic substances and nutrients may be due to emissions/discharges of wastewater from the Bucharest sewage system, as a result of the incapacity of the Glina treatment plant to take over the total volume of wastewater collected, some of the wastewater collected through the sewerage system being discharged into the Dambovită River. Also, along with this, some other possible pollution sources such as human agglomerations, discharges from economic agents, agricultural land leakage, uncontrolled waste deposits on the banks and riverbeds located upstream and downstream of the sewage treatment plant Glina can be considered.

In view of the data obtained for the water quality assessment of Dambovită River where high nutrient values, especially the nitrite concentrations were recorded, it can be concluded that the water quality of Dambovită River, in addition to those mentioned above, is influenced by the water quality of its tributary, the Colentina River which flows into it downstream of Bucharest.

This work provides basic information on the pollution status of the two rivers that are important sources as freshwater resources for residents of Bucharest and from the surrounding areas, therefore, frequent and systematic water quality monitoring is necessary in order to protect public health and improve the quality of life.

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